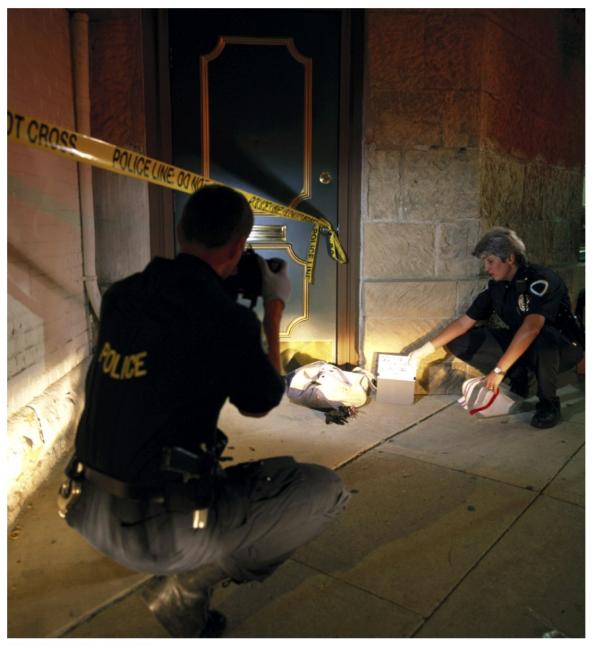
Collecting / Preserving Evidence

Chapter 2





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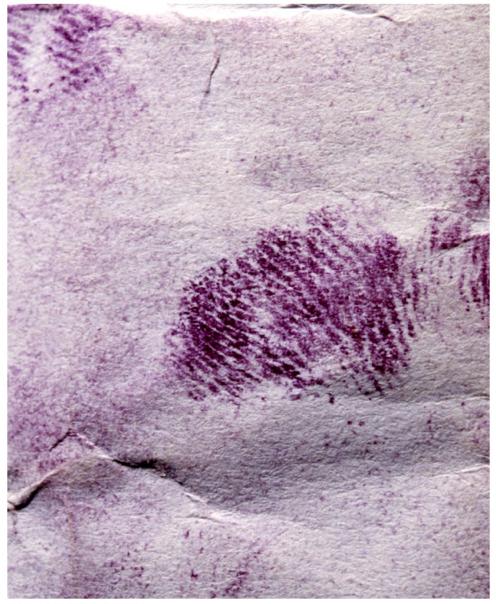
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Chemical Changes



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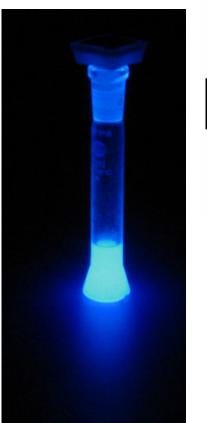


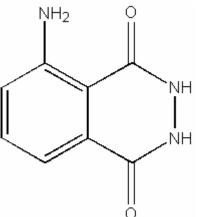




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Luminol







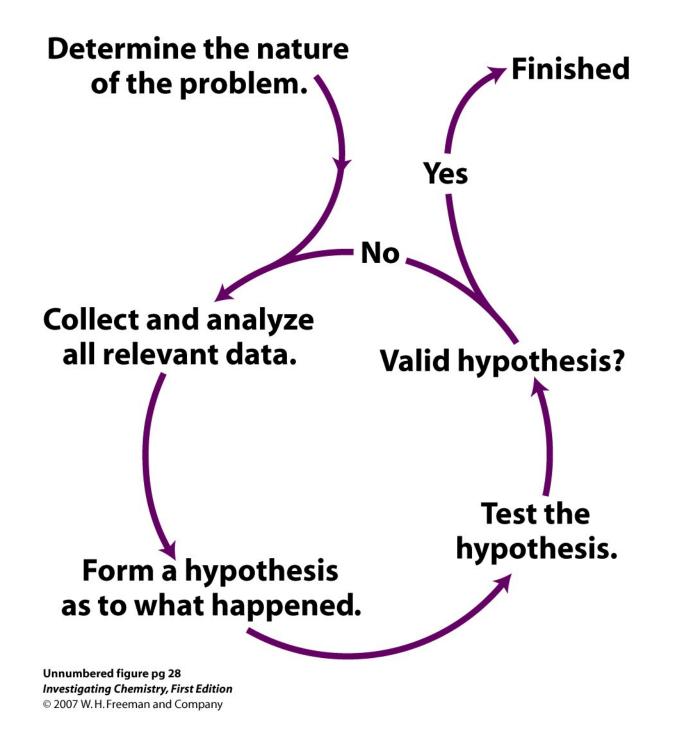






Physical Changes





Assignment: Scientific Method Find and read about the steps of the scientific method Watch video clip closely

Analyze the scene for

(1) Ways that Scientific Method is used

(2) Ways that Scientific Method is not used

(3) Based on (1) and (2), is the conclusion arrived at accurate?

Turn assignment in...

Table 2.1 Units of Measurement

Measurement Abbreviation	Units	
Mass	grams	g
Volume	liters	L
Distance	meters	m
Time	seconds	S

The standard units are used to measure experimental variables to facilitate communication between scientists.

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Table 2.2 Common Prefix Modifiers

Prefix	Abbreviation	Multiplier	
mega	Μ	1,000,000	
kilo	k	1,000	
deci	d	0.1	
centi	c	0.01	
milli	m	0.001	
micro	μ	0.000001	

Prefix modifiers are added to units in order to convey large or small quantities in a clear and concise manner.

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Table 2.3Physical Properties of Glass

Туре	Softening Point ¹ (°C)	Density (g/mL)	Refractive Index ²
Alkali barium	646	2.64	1.511
Alkali barium (optical)	647	2.60	1.512
Alkali barium borosilicate	712	2.27	1.484
Alkali borosilicate	718	2.29	1.486
Alkali strontium	688	2.26	1.519
Alkali zinc borosilicate	720	2.57	1.523
Borosilicate	720	2.28	1.490
Baria alumina borosilicate	844	2.76	1.530
Barium-alumina borosilicate	847	2.96	1.545
Borosilicate	821	2.23	1.473
Lanthanum barium	759	3.98	1.678
Lead borosilicate	447	5.46	1.860
Lead zinc borosilicate	370	3.80	_
Lithia potash borosilicate	_	2.13	1.469
Potash borosilicate	820	2.16	1.465
Potash soda lead	630	3.05	1.560
96% Silica	1530	2.18	1.458
96% Silica (porous)	1530	1.50	—
Silica (99.9% fused)	1585	2.20	1.459
Soda borosilicate	808	2.27	1.476
Soda alumina borosilicate	705	2.17	1.468
Soda-lime	696	2.47	1.510

¹The softening point is the temperature at which heated glass starts to deform under its own weight. ²The refractive index of all samples is measured at a wavelength of 589.3 nm.

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Soda borosilicate



borosilicate

Alkali strontium



Alkali barium borosilicate



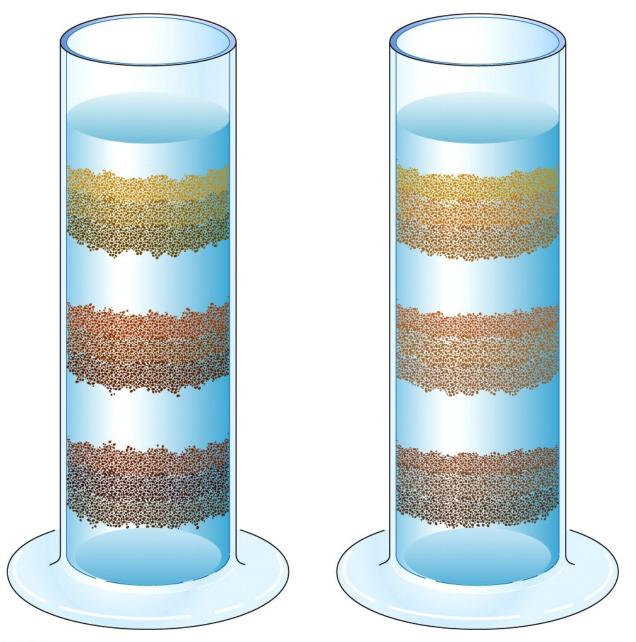


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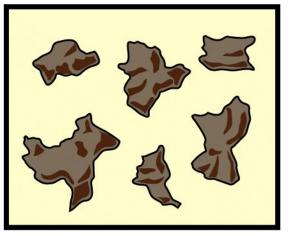
Granulation

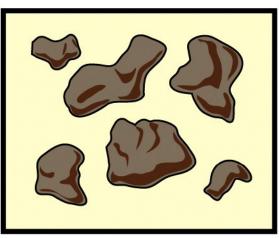
Boulders: Greater than 256 mm	
Cobbles: 64–256 mm	
Pebbles: 4–64 mm	
Granules: 2–4 mm	
Very coarse sand: 1–2 mm	
Coarse sand: 0.5–1 mm	
Medium sand: 0.25–0.5 mm	
Fine sand: 0.1–0.25 mm	
Very fine sand: 0.05–0.1 mm	
Silt: 0.002–0.05 mm	
Clay: Less than 0.002 mm	

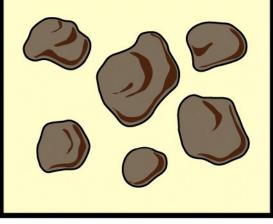
Figure 2-4a

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Rounding







Angular

Semirounded

Rounded

Figure 2-4b Investigating Chemistry, First Edition © 2007 W.H.Freeman and Company



Silicone dioxide, SiO₂ (quartz)

Calcium carbonate,

CaCO₃ (calcite)





Aluminum hydroxide, Al(OH), (gibbsite)

Figure 2-5c Investigating Chemistry, First Edition © 2007 W. H. Freeman and Company



Figure 2-5b Investigating Chemistry, First Edition © 2007 W.H. Freeman and Company

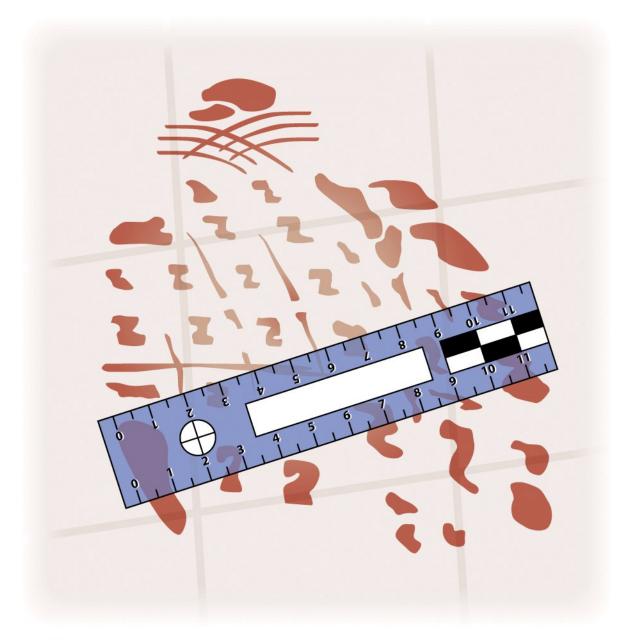
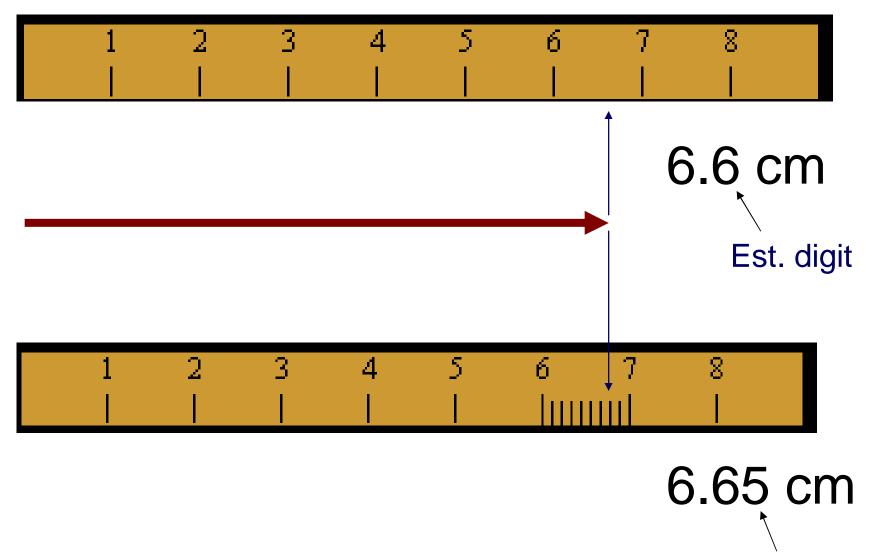


Figure 2-1 Investigating Chemistry, First Edition © 2007 W. H. Freeman and Company

Measurement and Accuracy



Est. digit



How pure is the cocaine??



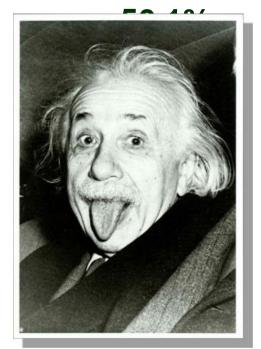


40.2% 53.6% 45.3%









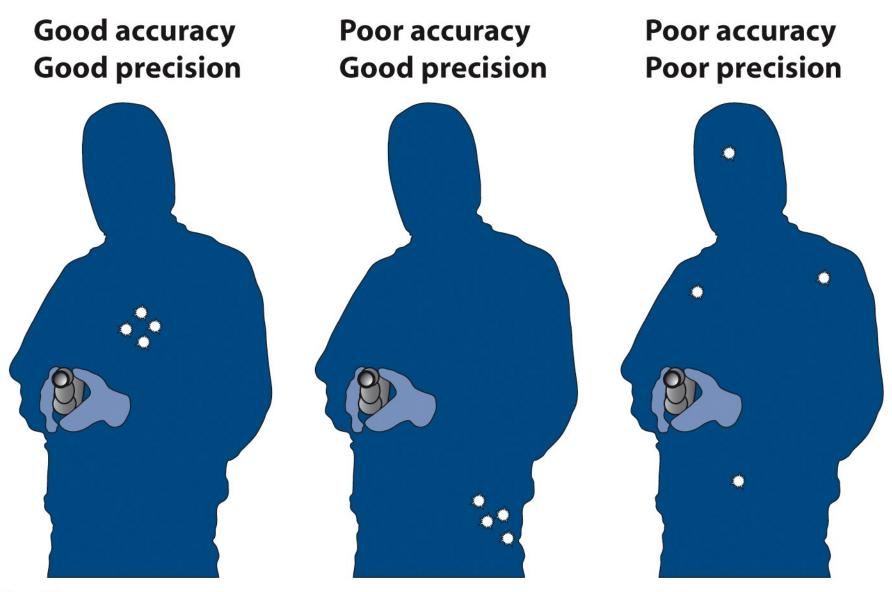


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