

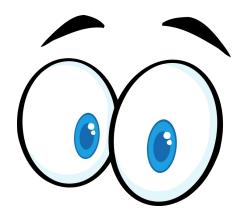


## Looking ahead

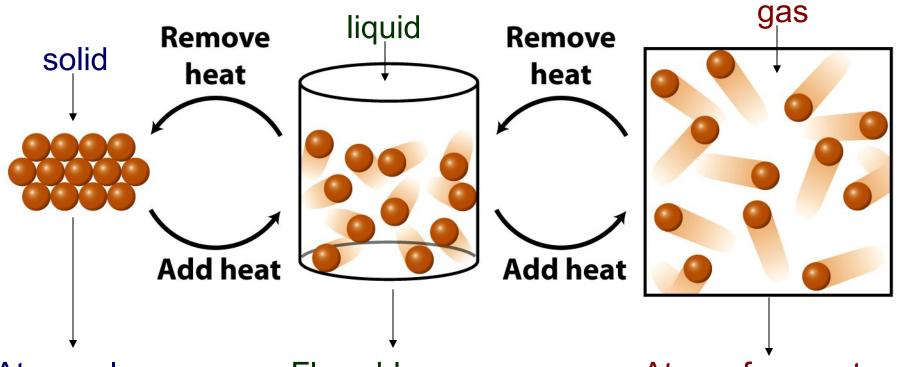
(to be completed before next lecture)

Chapter 1 problems: 3, 5-7, 38, 44, 46-47, 51-56

Read: Chapter 1.7-1.8, 1.12, 1.14, 6.1-6.2



#### States of Matter



Atoms close Fixed volume

**Flowable** Changes shape High IF "stickiness" Some IF "stickiness" Low kinetic energy Medium kinetic energy

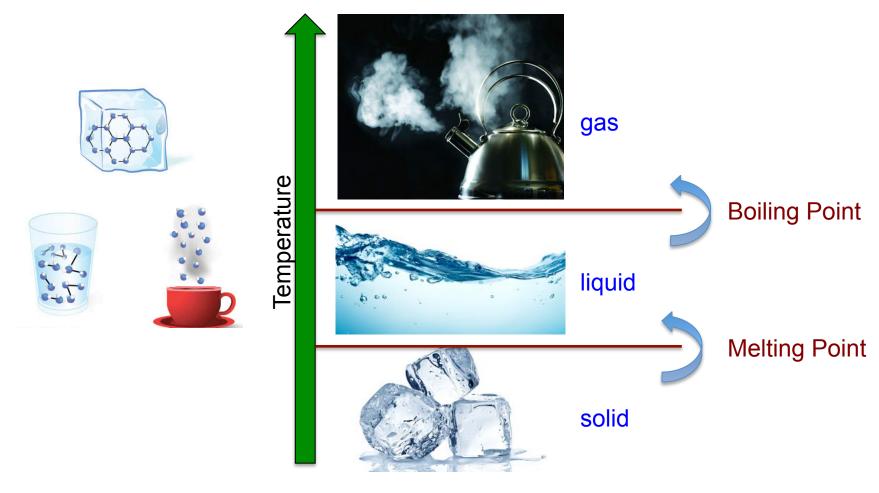
Atoms far apart Compressible Low IF "stickiness" High kinetic energy

#### States of Matter

How much do molecules in a sample like each other?

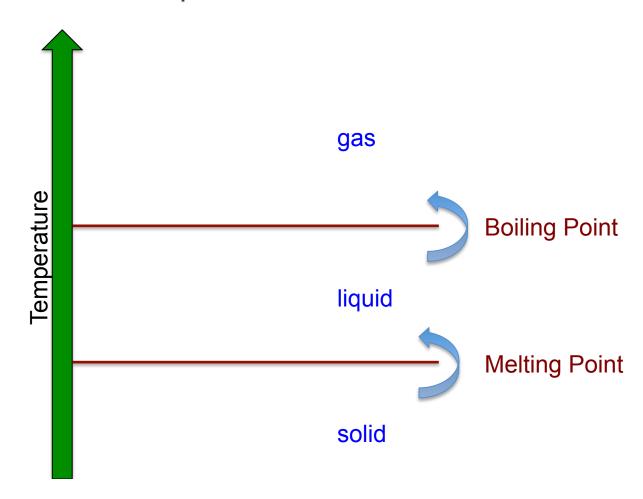
Boiling point: Temperature at which a liquid turns to gas

Melting point: Temperature at which a solid turns to liquid



#### Problem 1.2

Acetic acid, which gives the sour taste to vinegar, has a melting point of 16.7 °C and a boiling point of 118 °C. Predict the physical state of acetic acid when the ambient temperature is 10 °C.





## Types of Changes



Physical Change: A change that doesn't affect chemical makeup of a substance

Chemical Change: A change in the chemical composition of a substance



#### Problem 1.1

Identify each of the following as a physical change or a chemical change:

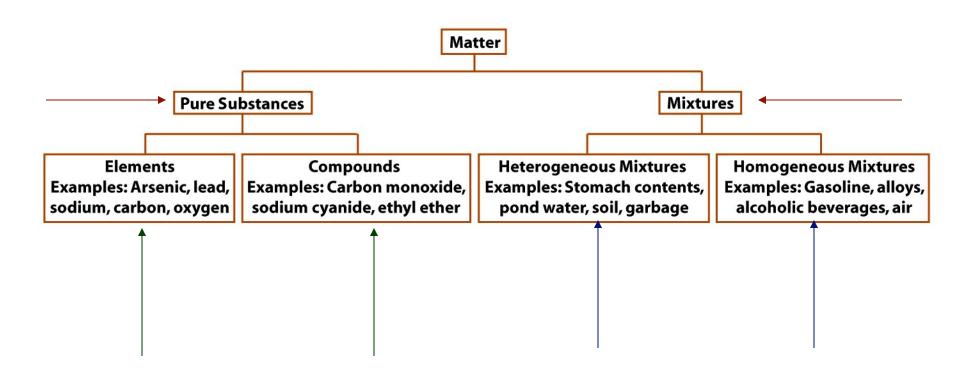
- a. Grinding of a metal
- b. Fruit ripening C
- c. Wood burning C
- d. A rain puddle evaporating

#### Problem 1.4

Classify each of the following as a physical change or a chemical change:

- a. Dissolving sugar in water
- b. Producing carbon dioxide gas and solid lime by heating C
   limestone
- c. Frying an egg
- d. The conversion of salicylic acid to acetylsalicylic acid (see the following Chemistry in Action)

## Types of Matter



## Homogeneous Mixtures



**Metal Alloys** 





Bullet



Figure 1-2b
Investigating Chemistry, First Edition
© 2007 W. H. Freeman and Company

Lighter Fluid

## Heterogeneous Mixtures



Investigating Chemistry, First Edition

© 2007 W. H. Freeman and Company



Figure 1-3c Investigating Chemistry, First Edition © 2007 W.H. Freeman and Company



Figure 1-3b Investigating Chemistry, First Edition © 2007 W. H. Freeman and Company

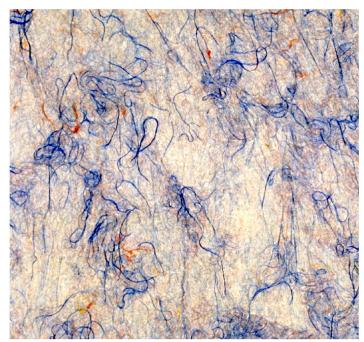


Figure 1-3d

Investigating Chemistry, First Edition
© 2007 W.H.Freeman and Company



## Healthy Booze???

Tannins are antioxidant molecules found in grape skin and seeds

Called Oligomeric Proanthocyanidins or OPC's

- -Help repair collagen and elastin
  - -Maintain blood vessels
  - -Prevent wrinkles
- -20X more powerful than Vit C and 50X than Vit E
- -Decrease histamine production

### Allergies and Antihistamines

The substances that cause an allergic reaction are called allergens or antigens

They bind to white blood cells called B-lymphocytes



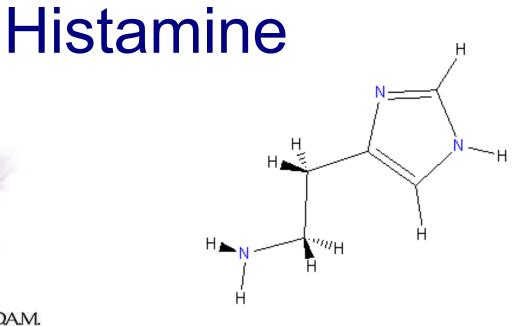
B-cells make Antibodies (Ab) against the allergen

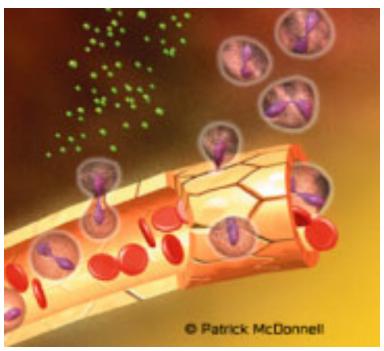


The Ab bind to mast cells and become receptors for the allergens

Next time the allergen enters the body, they bind to the Ab bound to the mast cell and cause histamine to be released.



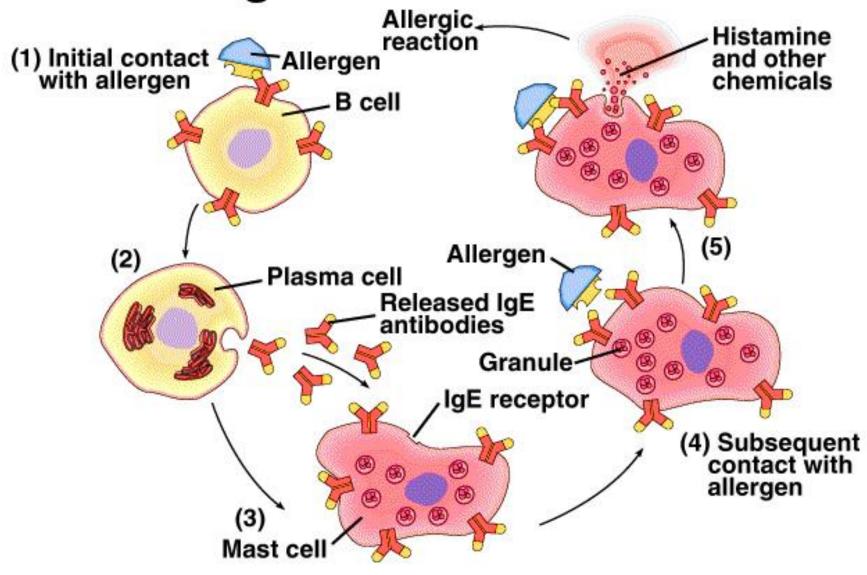




Mast Cell with histamine granules



## An Allergic Reaction — Overview



# OPC's Cont.

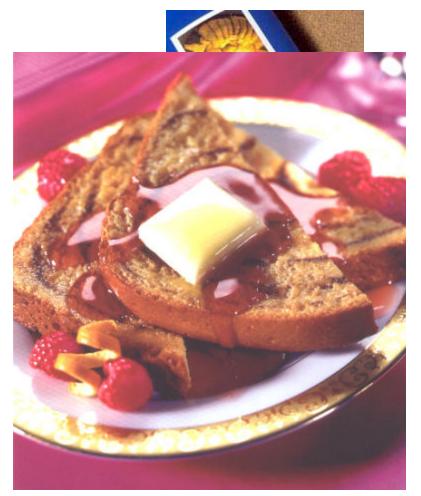
Fight smoke, alcohol, chemical damage ~~ Free Radicals

Internal sunscreen

Cross blood-brain barrier

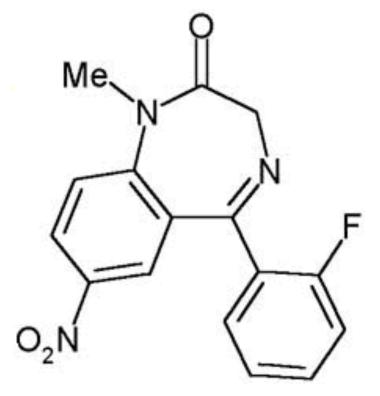
Explains French paradox







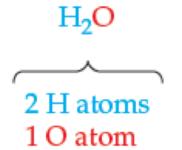
# Rohypnol



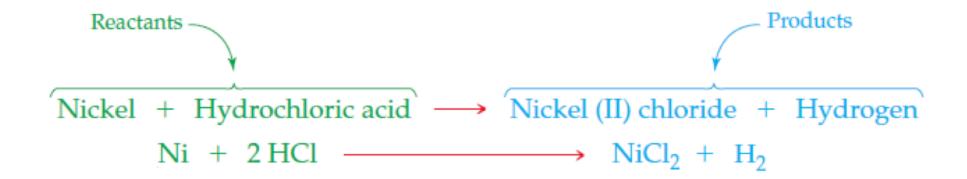


Rohypnol tablets are white and are single or cross-scored on one side with "ROCHE" and "1" or "2" encircled on the other.

(Graphics courtesy of the U.S Dept. of Justice)

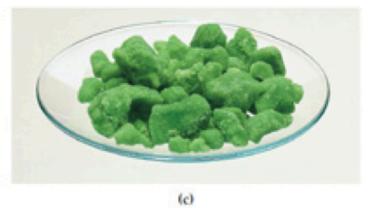






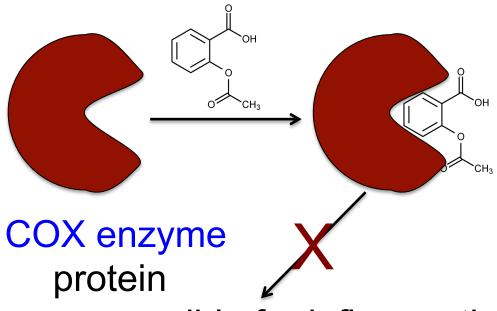








## **Aspirin**



responsible for inflammation helps with blood clotting makes stomach lining

