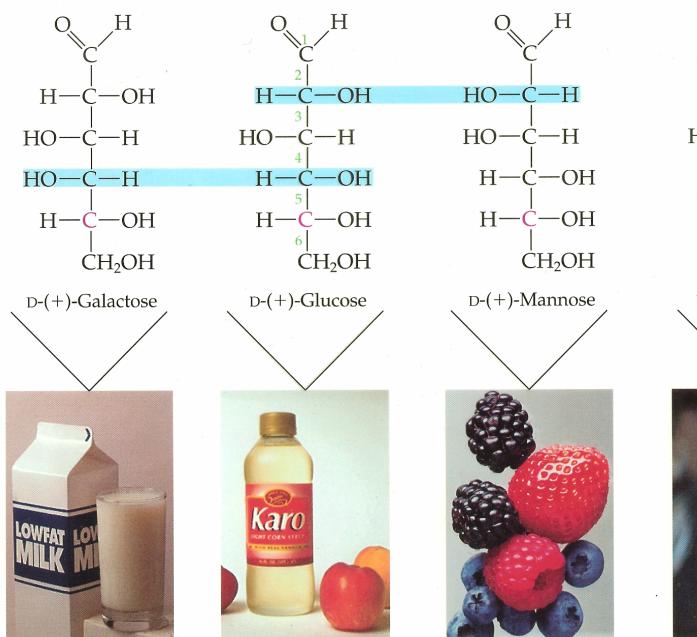
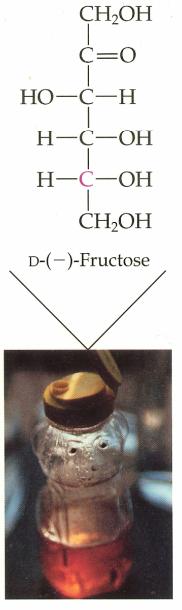
# Carbohydrates

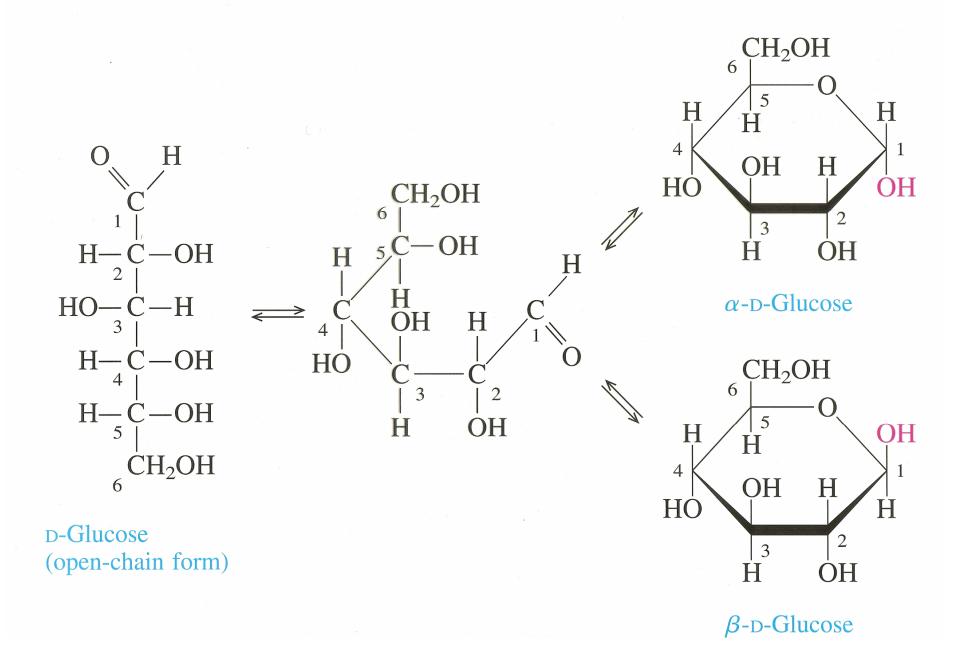
#### **Structures of Four Important Hexoses**

p. 509





## Mutarotation of Glucose



## Sugar (Glucose) Conformations

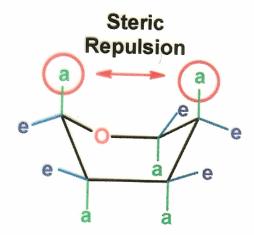


 $\beta$ -D-glucopyranose

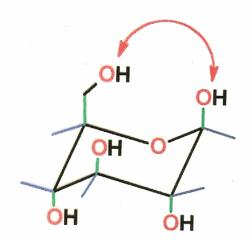
- -But cyclic form is NOT flat!!
- -Tetrahedral carbons, bent oxygen

**D-Glucose** (aldohexose)

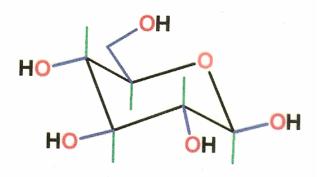
Chair conformation is predominate.



**β-D-glucopyranose** 



More steric repulsion



less steric repulsion

### **Two Possible Chair forms**

- (1) all substitutents (OH/CH2OH) equatorial (\*\*Most Stable!!\*\*)
- (2) all substituents (OH/CH2OH) axial (\*\*Less Stable\*\*)

Most stable conformation of Glucose



## **Cyclohexane Conformations**





-Not really flat -Tetrahedral Carbons

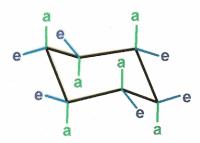
Cyclohexane







**Boat** Conformation

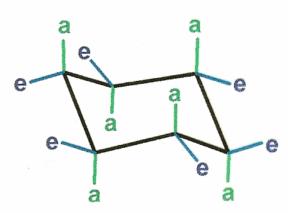


axial positions

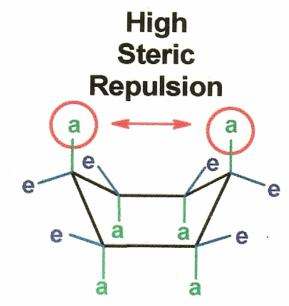
Perpendicular to Ring Plane

equatorial positions --- ~ in Ring Plane

## Chair Conformation is predominate form.



**Less Steric Repulsion** 



Level of steric repulsion between groups determines stability.

- High steric repulsion low stability
- Low steric repulsion high stability

Figure 19.4, p. 513

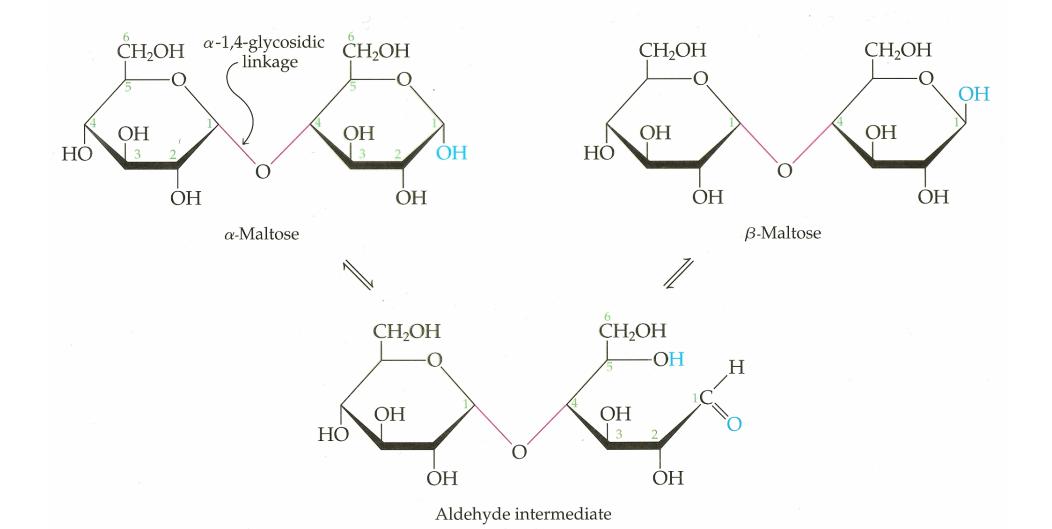
HOCH<sub>2</sub> O CH<sub>2</sub>OH 
$$\rightarrow$$
 HOCH<sub>2</sub> O OH  $\rightarrow$  OH  $\rightarrow$ 

## Mutarotation of Maltose

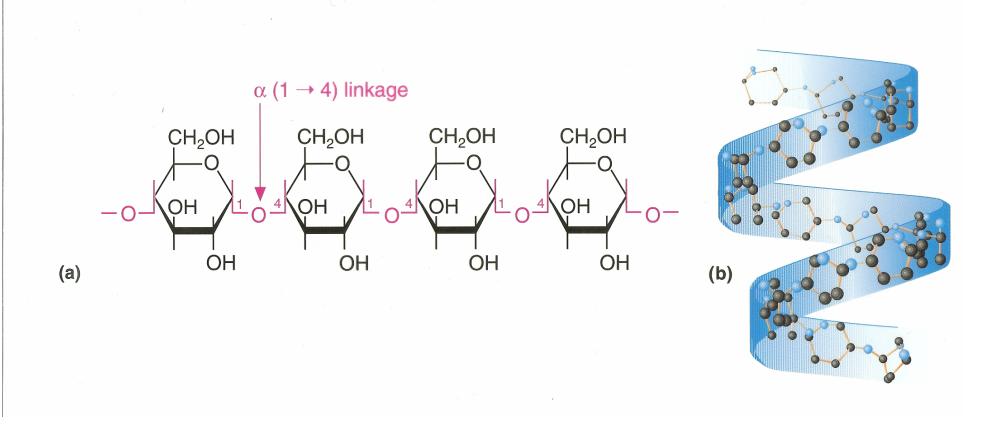
**T-66** 

Figure 19.6, p. 516

#### **Equilibrium Mixture of Maltose Isomers**

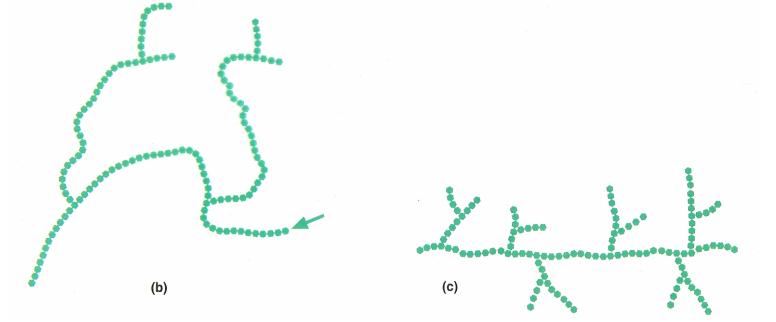


# Amylose



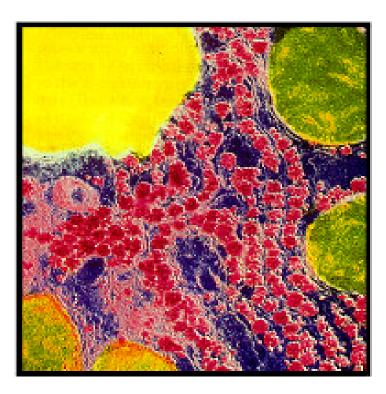
# Amylopectin

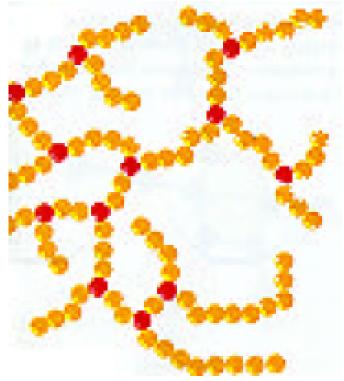
$$\begin{array}{c} \alpha \text{ (1 $\rightarrow$ 6) linkage} \\ -\text{O} \\ \text{OH} \\ \text{OH}$$



# **Glycogen** ЮH HO α-1,6 glycosidic bonds OH. o on ОН Н OH ŌН ŌН α-1,4 glycosidic bonds

## Glycogen forms long chains with many short branches.



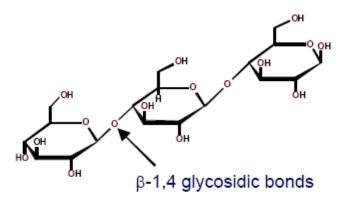


Pictures reproduced from Biochemistry, Voet/Voet

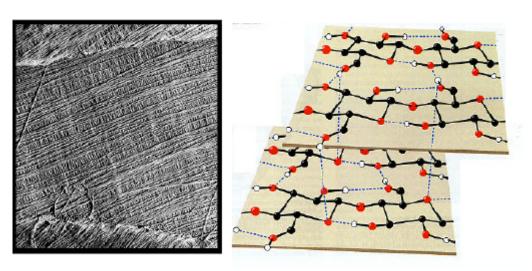


#### Cellulose





## Cellulose forms sheets because of extensive H-bonding between sugars.

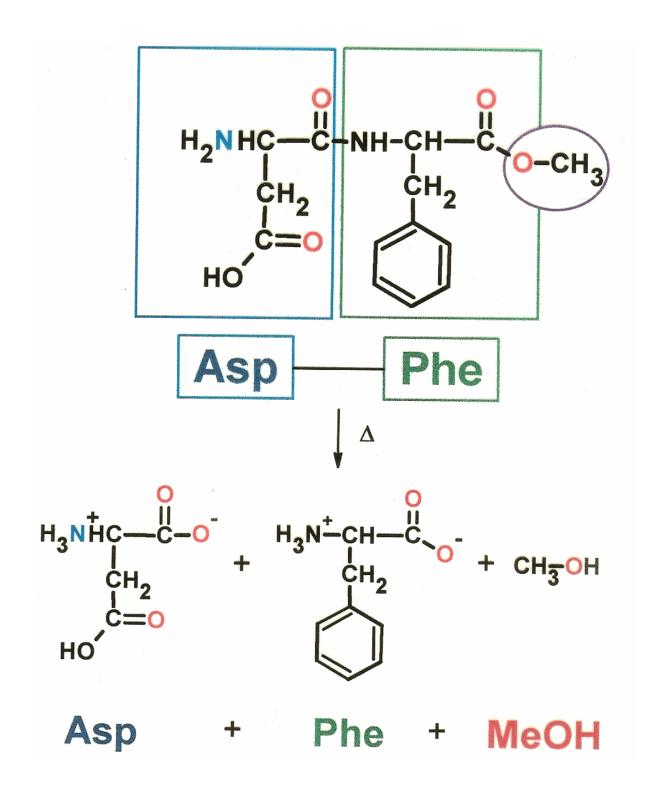


Pictures reproduced from Biochemistry, Voet/Voet

## <u>Aspartame</u>

Composed of 2 amino acids
Aspartic Acid (Asp)
Phenylalanine (Phe)

Asp tastes flat
Phe tastes bitter
But combined...
160X sweeter than sucrose





#### <u>Sucrose</u>





Non-reducing sugar

"head" to "head"

# Sodium Fluoride



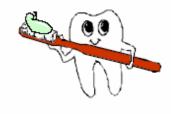
Used in tooth pastes to strengthen enamel and prevent decay

Used in blood collection tubes to preserve blood glucose and prevent ethanol production

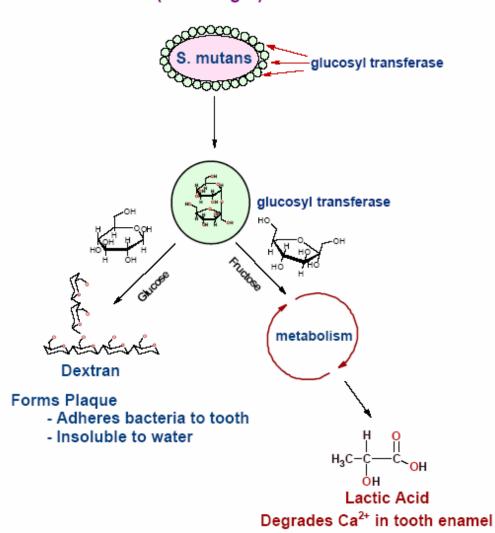
$$Na^+ + F^- \longrightarrow NaF$$



#### **Tooth Decay**



Streptococcus mutans cause of dental caries. Uses sucrose (table sugar).

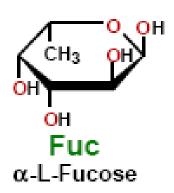


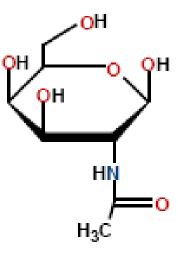




## **Blood Types**



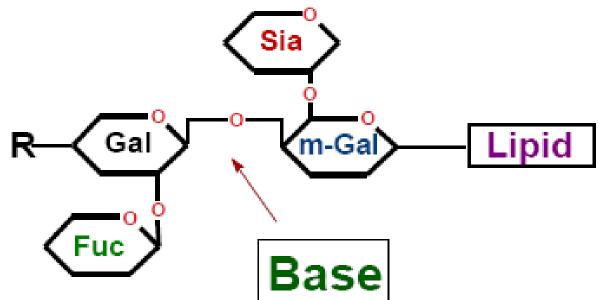


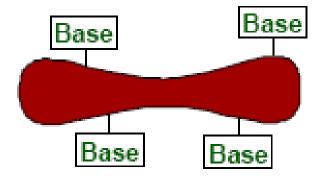


OH Sia

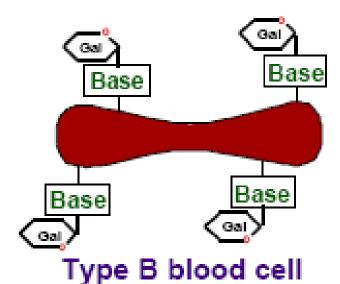
m-Gal β-D-N-Acetylgalactosamine

N-Acetylneuraminic acid



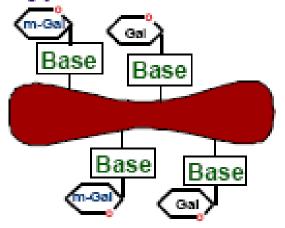


Type O blood cell



Base Base Base

Type A blood cell



Type AB blood cell

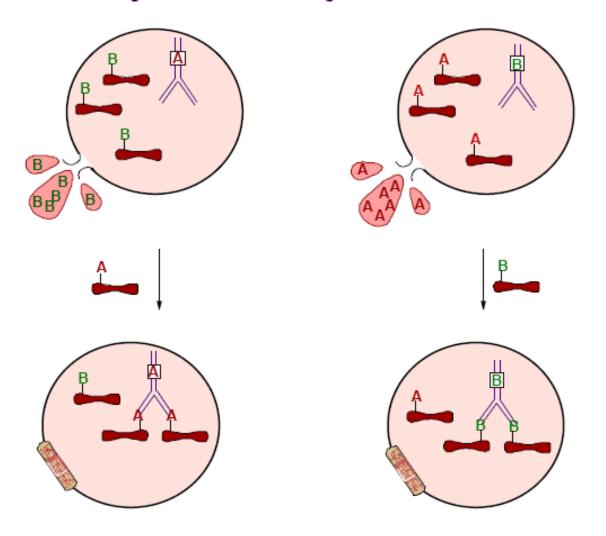


#### **Blood Types**

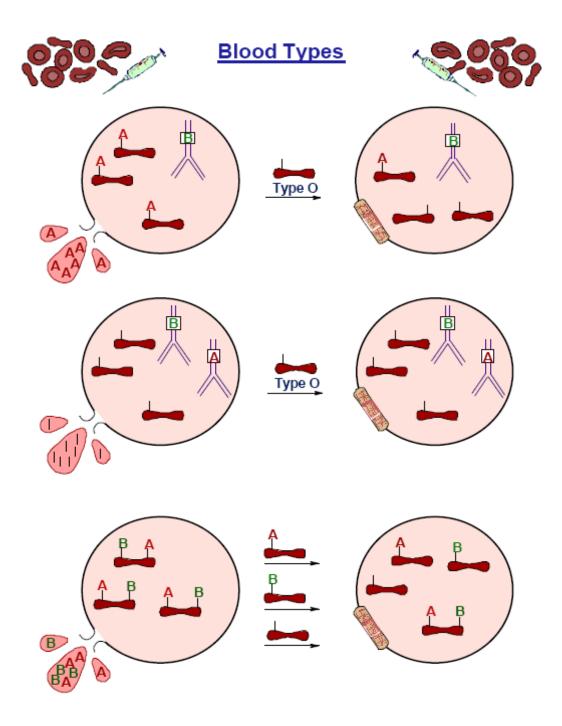


Need same type of blood because have antibodies to other R group.

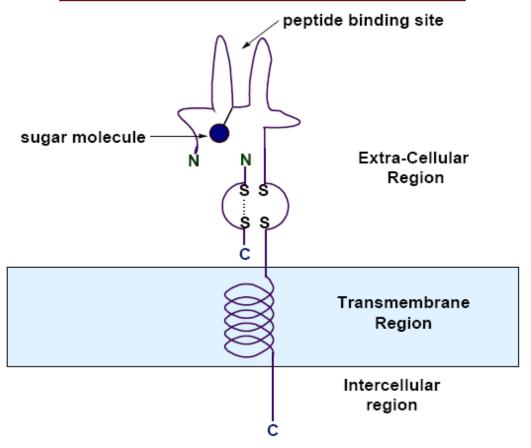
-A cant get B blood and B cant get A blood

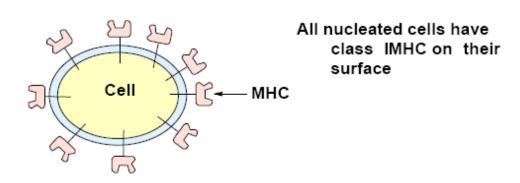


Antibodies bind to non-native RBC ultimately causing lysis.



#### Major Histocompatibility Complex (MHC)



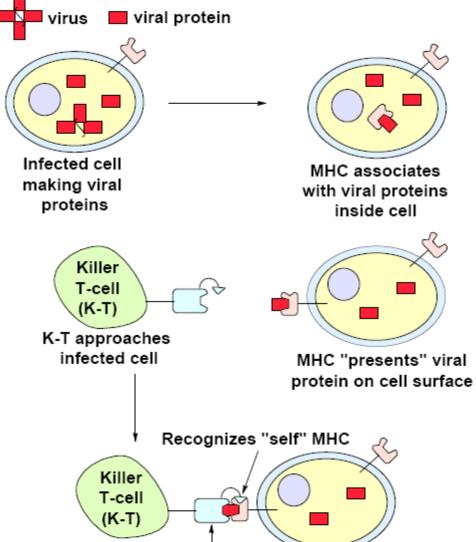




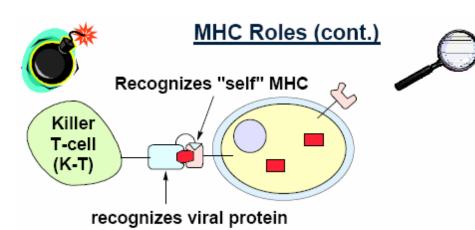
#### Roles of MHC

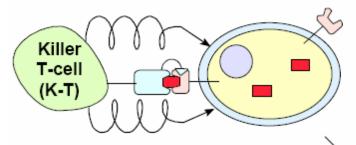


When cell infected by virus, viral proteins are produced inside



recognizes viral protein

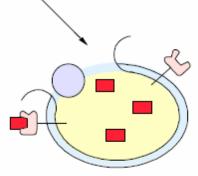




K-T then sends out chemical "bombs" that lyse virus infected cell!

So MHC:

- (1) presents viral antigens to K-T cells
- (2) provides self recognition
- (3) Reason that transplants are rejected.
- (4) Tissue Typing involves similarity between donor and recipient MHC.



Cell destroyed