FALL 2020 FINAL EXAM REVIEW

Comprehensive Review Material (about 40% of the final)

Numbers refer to problems from the current text.

Exam and quizzes numbers from those particular quizzes or exams.

Throughout the semester I've made comments about each of the exams.

It is worth your time to review these.

Periodicity, electron configurations

Examples 1.12,1.18,2.32, 2.27,2.28,2.31

Defining equations for IE and EA Qz1#6, Ex1#8

Excited vs ground state electron configurations Qz1#3, Ex1#4, Qz5#5, Ex3#4

Magnetism: para- vs dia- vs ferromagnetism

Orbital size, atom size Ex1#3, Qz1#4

Solids and common cubic packing arrangements

P, I, F packing patterns, 3 common cubic lattices common holes: location and number Ex2#3 cesium chloride, rock salt, zinc blende ionic lattices Ionic properties Qz3#2

Thermodynamics

Haber cycles Qz3#3, Ex2#4

standard energy of formation (meaning of little "f" Δ {G or H}_f°) Text 6.4, 6.6 Ex1#11c

K meaning and usage Qz2#2, Ex2#11,12

other thermo applications Text 6.13,18,22,34 Qz2#1,#3 Ex3#5,#6

Decomposition of polyatomic ion salts Qz4#3, Ex2#6a,#13-15

Redox

pH dependence oxidation number

oxidation numbers, balancing half reactions

n in balanced redox reactions

components of batteries and fuel cells Ex1#10, Ex2#8

electrolysis

electrolysis of water

reactive metals vs their metal ions

Hydridic/protic Hydrogen atoms

for example 15.25

Molecular orbital diagrams and chemistry they predict/explain

for example 3.3, 3.5, 3.6 Ex1#5-7, Ex1#14-16,Ex3#10

Mⁿ⁺(aq) acid base chemistry

for example contrast Fig 13.17 with behavior of B(OH)₃ Qz5#2,3

Al, Si, P structures common in nature 14.56, 14.45 See also p 344 Qz7#2-5, Ex3#8,9

Lewis dot/VSEPR 3.11, 3.12

Intermolecular forces 3.31 - 32

Lewis Acid/base Text 14.49, 7.33, 15.43 Ex3#2,3

New Material (about 60% of the final)

as from lecture, assigned reading, and practice homework

Oxygen family Ch 16 Halogen family Ch 17 Transition metal chemistry Ch 19