Give the name or formula for each of the following compounds. (make sure to include parenthesis where necessary)

1. Carbon tetrachloride
2. Magnesium acetate
3. $Hg(NO_3)_2$
4. Na_2SO_3
5. Zinc (II) hydroxide
6. $Sr(NO_3)_2$
7. HF
8. Iron (III) oxide
9. $Al(NO_3)_3$
10. $CaCO_3$
11. Sulfur dichloride
12. Magnesium phosphate
13. $BaSO_4$
14. Lithium chromate

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15. NaCN

16. Potassium chlorate

17. dihydrogen monoxide

18. $K_2Cr_2O_7$

19. SO_3 (not the charged polyatomic ion)

20. H_2S

21. Barium Hydroxide

22. Ammonium phosphate

23. Vanadium (V) sulfate

24. Lithium oxide

25. Na_2S

26. BiF_3

27. Cu_3P_2

28. NO

29. Lithium nitride

30. HBr

31. Chromium (III) carbonate

32. Ammonium acetate

33. Dinitrogen tetraoxide

34. CO_2

35. Potassium phosphate

36. $Pb(C_2H_3O_2)_2$

37. Mercury (I) chloride

38. $Sn(C_2H_3O_2)_4$

39. Magnesium chloride

40. $CaSO_4$

41. LiCl

42. $Cu_3(PO_4)_2$

43. $BaCrO_4$

44. $FeBr_2$

45. K_2O

46. Magnesium hydroxide

47. N_2O

48. H_2S

49. $CuSO_4$

50. $MgSO_4$

51. Sulfur difluoride

52. Tin (II) fluoride

53. Sodium hydride

54. RbBr

55. CsF

56. AlI_3

57. NF_3

58. N_2F_4

64. SiF_4

65. $Ru(NO_3)_3$

66. V_2O_5

67. Palladium (IV) sulfate

68. Iridium (III) nitride

69. TlCl

70. NiP

71. FeO

72. Titanium (II) sulfide

73. Titanium (IV) sulfate

74. CrO

75. Cs_2S

76. Rb_3P

77. Rubidium phosphate

78. Disulfur decafluoride

79. ClO_2 (not the charged polyatomic ion)

80. Boron trifluoride