

Give the name or formula for each of the following compounds. (make sure to include parenthesis where necessary)

1. Carbon tetrachloride

2. Magnesium acetate

3. $\text{Hg}(\text{NO}_3)_2$

4. Na_2SO_3

5. Zinc (II) hydroxide

6. $\text{Sr}(\text{NO}_3)_2$

7. HF

8. Iron (III) oxide

9. $\text{Al}(\text{NO}_3)_3$

10. CaCO_3

11. Sulfur dichloride

12. Magnesium phosphate

13. BaSO_4

14. Lithium chromate

15. NaCN

16. Potassium chlorate

17. dihydrogen monoxide

18. $K_2Cr_2O_7$

19. SO_3 (not the charged polyatomic ion)

20. H_2S

21. Barium Hydroxide

22. Ammonium phosphate

23. Vanadium (V) sulfate

24. Lithium oxide

25. Na_2S

26. BiF_3

27. Cu_3P_2

28. NO

29. Lithium nitride

30. HBr

31. Chromium (III) carbonate

32. Ammonium acetate

33. Dinitrogen tetraoxide

34. CO_2

35. Potassium phosphate

36. $\text{Pb}(\text{C}_2\text{H}_3\text{O}_2)_2$

37. Mercury (I) chloride

38. $\text{Sn}(\text{C}_2\text{H}_3\text{O}_2)_4$

39. Magnesium chloride

40. CaSO_4

41. LiCl

42. $\text{Cu}_3(\text{PO}_4)_2$

43. BaCrO_4

44. FeBr_2

45. K_2O

46. Magnesium hydroxide

47. N_2O

48. H_2S

49. CuSO_4

50. MgSO_4

51. Sulfur difluoride

52. Tin (II) fluoride

53. Sodium hydride

54. RbBr

55. CsF

56. AlI_3

57. NF_3

58. N_2F_4

64. SiF_4

65. $\text{Ru}(\text{NO}_3)_3$

66. V_2O_5

67. Palladium (IV) sulfate

68. Iridium (III) nitride

69. TlCl

70. NiP

71. FeO

72. Titanium (II) sulfide

73. Titanium (IV) sulfate

74. CrO

75. Cs_2S

76. Rb_3P

77. Rubidium phosphate

78. Disulfur decafluoride

79. ClO_2 (not the charged polyatomic ion)

80. Boron trifluoride