Measurement and Significant Figures

Tro: Chemistry: A Molecular Approach, 2/e

What Is a Measurement?

- Quantitative observation
- Comparison to an agreed standard
- Every measurement has a number and a unit Meniscus

A Measurement

- The unit tells you what standard you are comparing your object to
- The number tells you
 - 1. what multiple of the standard the object measures
 - 2. the uncertainty in the measurement
- Scientific measurements are reported so that every digit written is certain, except the last one, which is estimated

Estimating the Last Digit

- For instruments marked with a scale, you get the last digit by estimating between the marks
- Mentally divide the space into ten equal spaces, then estimate how many spaces over the indicator the mark is.

Estimation in Weighing





(b)

Markings every 0.1 g Estimated reading 1.27 g

Markings every 1 g Estimated reading 1.2 g

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Estimating the last digit



•For the scale to the left, the smallest marked division is 0.2.

•Divide that by 10, so you will estimate to 0.02, 0.04,...

•How would you report the measurement?

Digital Readouts

- These are easy!
- Report ALL digits you see, the instrument has already done the rounding for you.

Precision and Accuracy

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Uncertainty in Measured Numbers

- Sources of error:
 - the experimental design/experimenter
 - Faulty or uncalibrated instruments
 - nature's random behavior
- Accuracy is an indication of how close a measurement comes to the actual value of the quantity
- Precision is an indication of how close repeated measurements are to each other
 - how reproducible a measurement is

Precision

Do you get about the same value each time you measure?

- We determine the precision by evaluating how far the measurements are from each other
- You may miss the correct measurement, but you are consistent

Accuracy

How "right" is my measurement?

• We determine the accuracy of a measurement by evaluating how far it is from the actual value

Types of Errors

- Inaccuracy in measurement caused by systematic errors
 - errors we can fix by using more accurate instruments, or better technique or experimental design
- Imprecision in measurements is caused by random errors
 - errors that result from random fluctuations
 - no specific cause, therefore cannot be corrected