

# Key

Draw Lewis Dot Structures for the following molecules:

1. CO



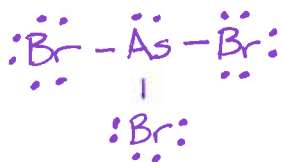
6. SCl<sub>2</sub>



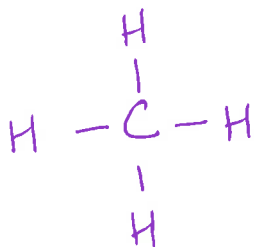
2. CH<sub>2</sub>O



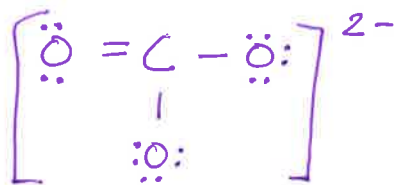
7. AsBr<sub>3</sub>



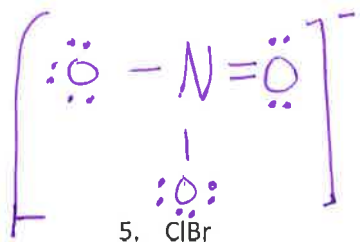
3. CH<sub>4</sub>



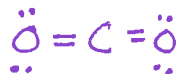
8. CO<sub>3</sub><sup>2-</sup>



4. NO<sub>3</sub><sup>-</sup>



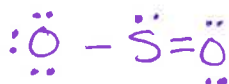
9. CO<sub>2</sub>



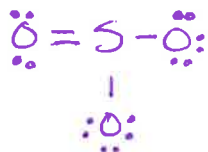
5. ClBr



10. SO<sub>2</sub>



11.  $\text{SO}_3$



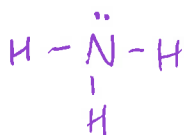
12.  $\text{IBr}$



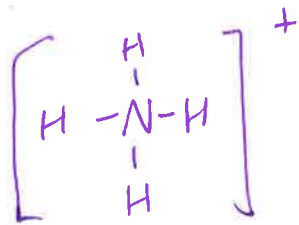
13.  $\text{SiO}_2$



14.  $\text{NH}_3$



15.  $\text{NH}_4^+$



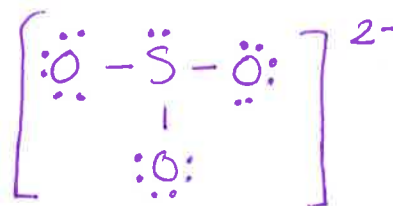
16.  $\text{SeO}_2$



17.  $\text{HCN}$



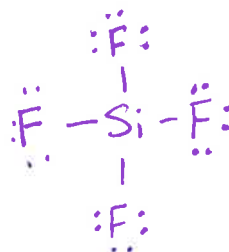
18.  $\text{SO}_3^{2-}$



19.  $\text{H}_2\text{O}$



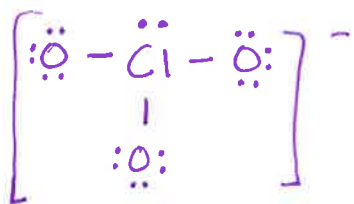
20.  $\text{SiF}_4$



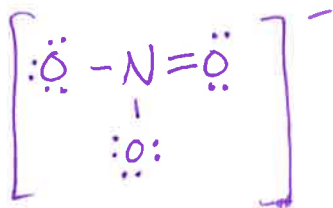
21.  $\text{NF}_3$



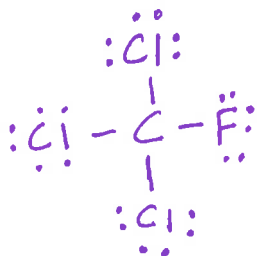
22.  $\text{ClO}_3^-$



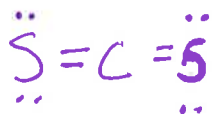
23.  $\text{NO}_3^-$



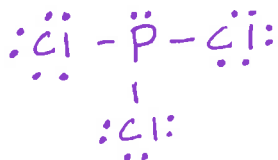
24.  $\text{CFCl}_3$



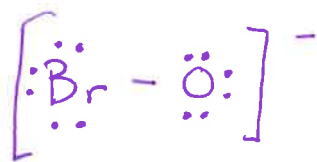
25.  $\text{CS}_2$



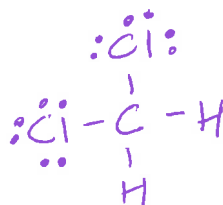
26.  $\text{PCl}_3$



27.  $\text{BrO}^-$



28.  $\text{CH}_2\text{Cl}_2$



29.  $\text{Br}_2\text{O}$



30.  $\text{H}_2\text{S}$

