

NAME _____

Physiological Chemistry II

Final

Dr. Melissa Kelley

April 27, 2004

Provide the one best answer for each, following the instructions given in each section of the exam. You have 2 hours to complete the exam.

Match the functional group listed below with the compounds shown below. Questions 1a-f.

Anhydride

Ketone

Aldehyde

Amide

Carboxylic acid

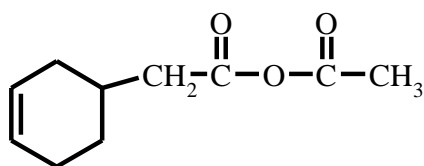
Ester

Alcohol

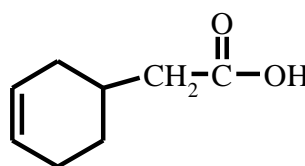
Ether

Amine

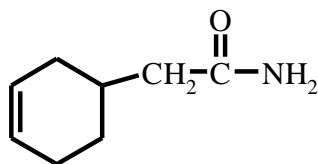
Lactone



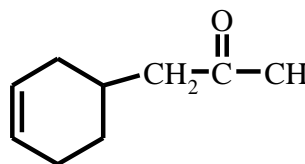
a. _____



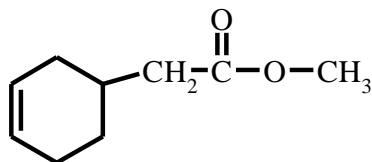
b. _____



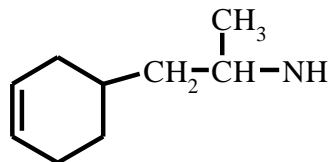
c. _____



d. _____



e. _____



f. _____

Multiple Choice: Select the one best answer to each question. Questions 2-39.

_____ 2. Which of the following amino acids would likely be located on the exterior of a protein?

- A. Aspartate
- B. Isoleucine
- C. Phenylalanine
- D. Tryptophan

_____ 3. Addition of water to an alkene will yield which of the following products?

- A. Aldehyde
- B. Ketone
- C. Anhydride
- D. Alcohol

- _____ 4. Which of the following compounds is an intermediate in the TCA cycle?
- Oxaloacetate
 - Aspartate
 - Ornithine
 - Citrulline
- _____ 5. Which of the following pathways is used to produce ribose?
- Glycolysis
 - Gluconeogenesis
 - Pentose phosphate pathway
 - β -oxidation
- _____ 6. Which of the following amino acids would be involved in a salt bridge of a protein?
- Serine
 - Cysteine
 - Methionine
 - Lysine
- _____ 7. Oxidation of a secondary alcohol will yield which of the following products?
- Amide
 - Ketone
 - Ester
 - Ether
- _____ 8. Which of the following lipids would not be found in animal cell membranes?
- Phospholipids
 - Triacylglycerides
 - Sphingolipids
 - Cholesterol
- _____ 9. Which of the following compound is a base?
- $$\text{H}_3\text{C}-\text{CH}_2-\overset{\text{O}}{\parallel}{\text{C}}-\text{NH}_2$$
 - $$\text{H}_3\text{C}-\text{CH}_2-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$$
 - $$\text{H}_3\text{C}-\text{CH}_2-\text{CH}_2-\text{NH}_2$$
 - $$\text{H}_3\text{C}-\text{CH}_2-\text{CH}_2-\text{OH}$$
- _____ 10. Which of the following is the central dogma of molecular biology?
- DNA? RNA? Protein? DNA
 - DNA? DNA? RNA? Protein
 - RNA? Protein? DNA? DNA
 - Protein? DNA? RNA? DNA

- _____ 11. Which of the following statements about enzymes is **not correct**?
- A. Enzymes do not alter the equilibrium of a reaction.
 - B. Enzymes are reused and regenerated.
 - C. Enzymes increase the energy barrier of a reaction.
 - D. Enzymes accelerate the rate of the reaction.
- _____ 12. Which of the following compounds is **not** an intermediate of the urea cycle?
- A. α -ketoglutarate
 - B. Citrulline
 - C. Ornithine
 - D. Arginine
- _____ 13. Which of the following is **not true** of electron transport and oxidative phosphorylation?
- A. NAD and FAD from the TCA cycle are used to generate ATP in the electron transport chain.
 - B. Electrons are passed through protein complexes containing cytochromes.
 - C. Protons are pumped from the mitochondrial matrix to the intermembrane space of the mitochondria.
 - D. Oxidative phosphorylation involves formation of ATP.
- _____ 14. Anhydrides react with amines to form which of the following products?
- A. Carboxylic acid and an ester
 - B. Carboxylic acid and an amide
 - C. Alcohol and an amide
 - D. Alcohol and an ester
- _____ 15. Ferritin is an iron storage protein, which has a bundle of α -helices that it uses to bind iron. Which of the following levels of protein structure best defines these α -helices?
- A. Primary structure
 - B. Secondary structure
 - C. Tertiary structure
 - D. Quaternary structure
- _____ 16. Which of the following is **not true** of glucose?
- A. It is an aldohexose.
 - B. It is the end product of gluconeogenesis.
 - C. It is a reducing sugar.
 - D. It can be used to form ketone bodies.
- _____ 17. Hydrolysis of an ester will yield which of the following products?
- A. Carboxylic acid and an alcohol
 - B. Carboxylic acid and an ether
 - C. Amide and an ether
 - D. Anhydride and an amine

- _____ 18. Which of the following best describes glycogen?
- A. A polymer of ribose residues joined by *beta*-1,4-glycosidic bonds with 1,6 branches.
 - B. A polymer of ribose residues joined by *alpha*-1,4-glycosidic bonds with 1,6 branches.
 - C. A polymer of glucose residues joined by *beta*-1,4-glycosidic bonds with 1,6 branches.
 - D. A polymer of glucose residues joined by *alpha*-1,4-glycosidic bonds with 1,6 branches.
- _____ 19. Which of the following statements is **not true**?
- A. Acetyl-CoA is a thioester.
 - B. Ketone bodies are produced from Acetyl-CoA.
 - C. All carbons in cholesterol are from Acetyl-CoA.
 - D. The end product of lipogenesis is Acetyl-CoA.
- _____ 20. Which of the following hormones stimulates lipogenesis?
- A. Insulin
 - B. Glucagon
 - C. Prostaglandins
 - D. Vitamin A
- _____ 21. The bond that connects the base to the sugar in DNA or RNA is:
- A. A phosphate ester bond.
 - B. A phosphate anhydride bond.
 - C. A N-glycosidic bond.
 - D. A phosphate nitrogen bond.
- _____ 22. Arachidonic acid serves as a precursor to which of the following compounds?
- A. Cholesterol
 - B. Acetyl-CoA
 - C. Leukotrienes
 - D. Oxaloacetate
- _____ 23. The carbons of the amino acid cysteine are metabolized to pyruvate. Which of the following best describes cysteine?
- A. Cysteine is a ketogenic amino acid.
 - B. Cysteine is a glucogenic amino acid.
 - C. Cysteine is both a glucogenic and ketogenic acid.
 - D. Cysteine is neither a glucogenic or ketogenic amino acid.
- _____ 24. Which of the following statements is **not correct**?
- A. TAG mobilization from adipocytes occurs when insulin levels are high.
 - B. TAG storage in adipocytes requires glycerol-3-phosphate.
 - C. Fatty acids are esterified to glycerol-3-phosphate.
 - D. TAG mobilization results in many fatty acids released into the blood stream bound to serum albumin.

- _____ 25. Which of the following amino acids serves as a carrier of amino groups into the urea cycle?
- A. Aspartate
 - B. Citrulline
 - C. Glutamate
 - D. Ornithine
- _____ 26. In the 1950's Watson and Crick proposed the DNA double helix as a model for DNA structure. Which of the following statements concerning the DNA double helix is **not correct**?
- A. The bases are located on the outside of the helix.
 - B. The bases are capable of hydrogen bonding.
 - C. The two strands are antiparallel.
 - D. The DNA molecule consists of two polynucleotide strands.
- _____ 27. In lipogenesis, Acetyl-CoA does not cross the mitochondrial membrane but condenses with oxaloacetate to form which of the following compounds that crosses the mitochondrial membrane?
- A. Acetoacetate
 - B. Citrate
 - C. Aspartate
 - D. β -hydroxybuterate
- _____ 28. In lipogenesis, the carbons required for a fatty acid to elongate come from which of the following sources?
- A. Acetoacetate
 - B. Malate
 - C. Pyruvate
 - D. Malonyl-CoA
- _____ 29. Which of the following lipids contain sphingosine as a backbone?
- A. Phospholipids
 - B. TAG
 - C. Cholesterol
 - D. Glycolipids
- _____ 30. Which of the following is **not true** of ketone body formation?
- A. During starvation β -oxidation of fatty acids leads to decrease in Acetyl-CoA and decrease in ketone body formation.
 - B. Acetoacetate, β -hydroxybuterate and acetone are ketone bodies.
 - C. Ketone bodies are carboxylic acids.
 - D. Some amino acids can make ketone bodies.
- _____ 31. Which of the following is **true** of DNA?
- A. DNA contains the bases A, G, C, and T and ribose as the sugar.
 - B. DNA contains the bases A, G, C, and T and deoxyribose as the sugar.
 - C. DNA contains the bases A, G, C, and U and ribose as the sugar.
 - D. DNA contains the bases A, G, C, and U and deoxyribose as the sugar.

- _____ 32. Which of the following compounds is the end product of β -oxidation?
- A. Malonyl-CoA
 - B. Oxaloacetate
 - C. Acetyl-CoA
 - D. Glutamate
- _____ 33. Which of the following serves as the five carbon unit that undergoes condensation reactions in cholesterol synthesis?
- A. Isoprenes
 - B. Arachidonic acid
 - C. Leukotrienes
 - D. Malonyl-CoA
- _____ 34. Biosynthesis of glutamate requires which of the following as a carbons source?
- A. Oxaloacetate
 - B. α -ketogluterate
 - C. Pyruvate
 - D. Aspartate
- _____ 35. Long chain unsaturated fatty acids can be converted to saturated fatty acids by which of the following reactions?
- A. Transamination
 - B. Oxidative deamination
 - C. Hydrogenation
 - D. Claisen condensation
- _____ 36. These sugars are usually found attached to proteins that are responsible for blood typing:
- A. Monosaccharides
 - B. Disaccharides
 - C. Oligosaccharides
 - D. Polysaccharides
- _____ 37. Which of the following statements about phospholipids is **incorrect**?
- A. They contain a phosphoanhydride bond.
 - B. They contain glycerol as the backbone.
 - C. They contain two fatty acids
 - D. They contain a polar phosphate group.
- _____ 38. Which of the following is **not true** of β -oxidation?
- A. FADH_2 is produced.
 - B. NADH is produced.
 - C. β -oxidation is stimulated in response to glucagon.
 - D. β -oxidation is stimulated in response to insulin.

_____39. This compound is formed in the liver from Acetyl-CoA during starvation as a source of carbon units for brain metabolism.

- A. Acetoacetate
- B. Cholesterol
- C. Malonyl-CoA
- D. Glutamate

Associate the Following Pathways with the Reactions Presented. Questions 40-46.

_____40. Pentose Phosphate Pathway

_____41. Glycolysis

_____42. Urea cycle

_____43. Gluconeogenesis

_____44. β -oxidation

_____45. Lipogenesis

_____46. Glycogen synthesis

A. Palmitic acid ? 8 Acetyl-CoA + 7 NADH + 7 FADH₂

B. 2 Lactate + 6 ATP ? Glucose

C. Glycogen + Pi ? Glucose-1-phosphate

D. NH₄⁺ + CO₂ + 2 ATP ? Carbamoyl phosphate

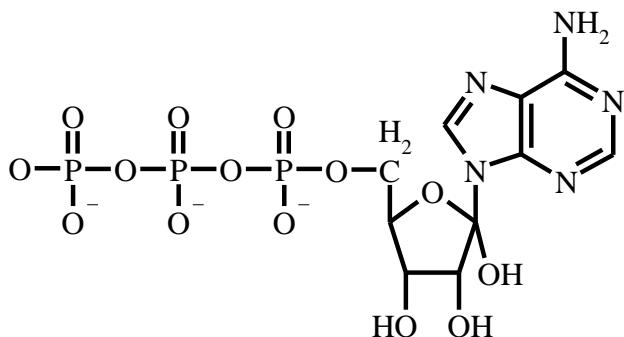
E. Glucose ? 2 Pyruvate + 2 ATP

F. Glucose-6-P + 2 NADP ? Ribose-5-P + 2 NADPH

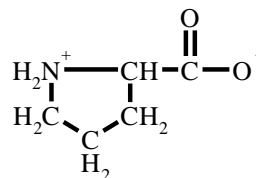
G. Glucose-1-P + UTP ? UDP-glucose + PPi

H. 7 Acetyl-CoA + 14 NADPH ? Myristic acid

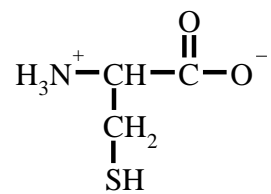
Name or identify the compounds shown below. Questions 47-61.



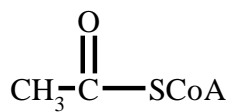
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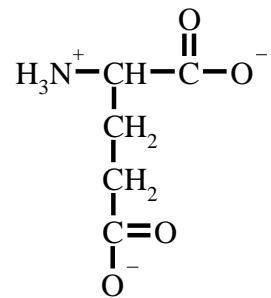
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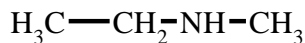
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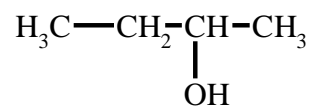
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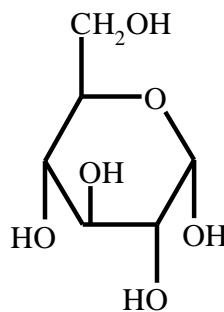
51. _____



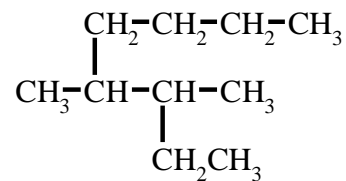
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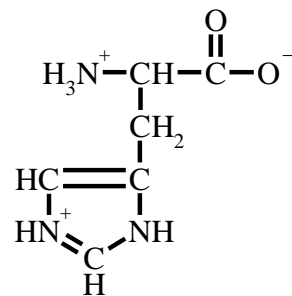
53. _____



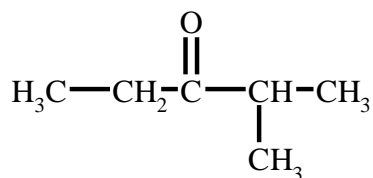
54. _____



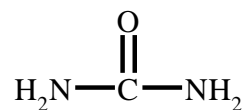
55. _____



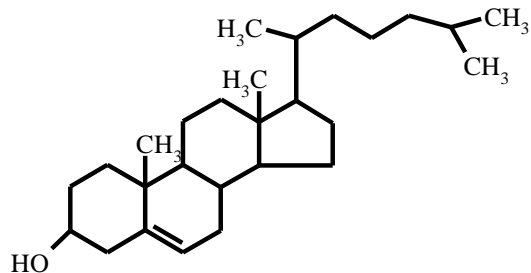
56. _____



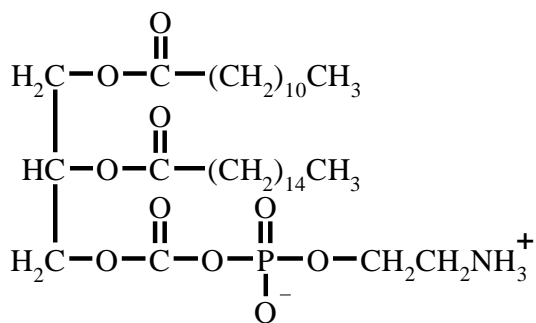
57. _____



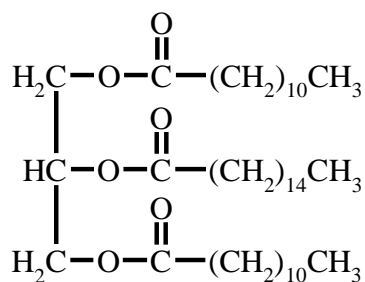
58. _____



59. _____



60. _____

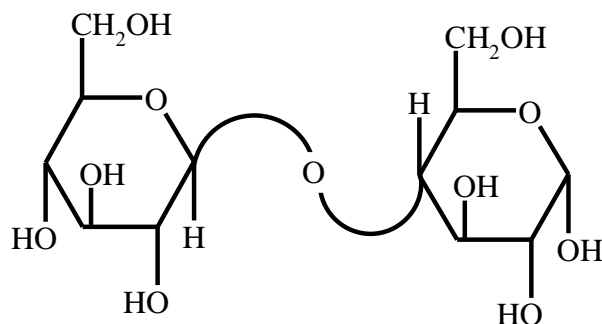


61. _____

Short answer, fill in the blank (giving one or two word responses). If multiple answers are correct, any one answer is acceptable. Questions 62-75

62. Given the DNA sequence below, write the complementary sequence of the second strand of DNA.
5'-AGCTAGGTT-3'

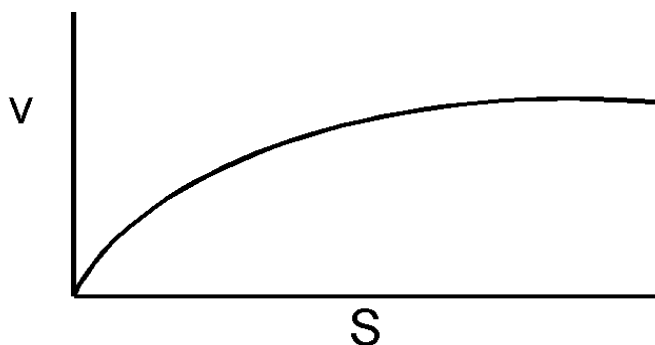
63. (3 points) Shown below is a disaccharide. Identify if this a reducing or non-reducing sugar. Circle the anomeric carbon. Place a square around all the atoms involved in the glycosidic linkage.



_____ 64. This vitamin serves as a precursor to the cofactors NAD, NADP, NADH, and NADPH. Name this vitamin.

65. In the space provided below, draw the structure of 3-propylcyclohexene.

66. In the chart below, the change in reaction rate (v) versus increasing substrate concentration for an uninhibited enzyme is shown. Draw the curve expected if a non-competitive inhibitor is added to the reaction.



_____ 67. This vitamin serves as a precursor to the cofactors FAD, FMN, and FADH₂. Name this vitamin.

68. In the space provided below draw the structure of β -D-glucose.

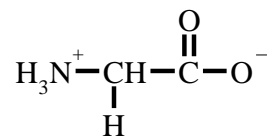
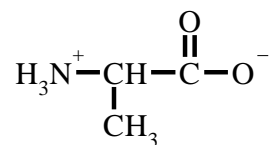
_____69. Nucleic acids are long polymers of nucleotides connected by what type of bond?

_____70. This vitamin is essential in vision and growth promoting activity. Name the vitamin.

71. In the space provided below draw the structure of 2-ethylheptanal.

72. In the space provided below draw the structure of α -D-ribose.

73. (4 points) Shown below are alanine and glycine. Draw the dipeptide that would form if glycine is the N-terminal amino acid and alanine is the C-terminal amino acid. Place a square around the atoms involved in the peptide bond.

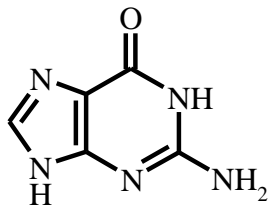


74. (5 points) In this class we talked about how Acetyl-CoA is made through metabolism. Name two compounds that acetyl-CoA serves as a precursor to:

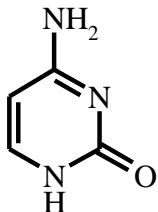
1.

2.

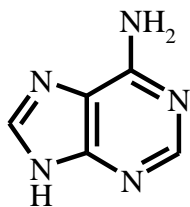
75. (4 points each) Shown below are the bases. Name the base, identify whether the base is a purine or pyrimidine, and whether the base is found in DNA, RNA or both.



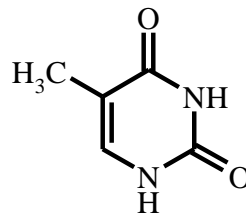
a. _____



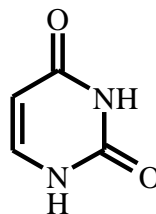
c. _____



e. _____



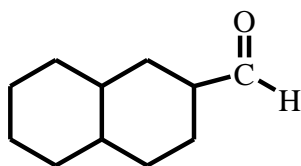
b. _____



d. _____

In the following questions, provide the product of the reaction given. If there is no reaction, specify no reaction (NR) in the answer. Questions 76-83.

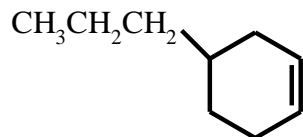
76.



Mild oxidation
 \longrightarrow



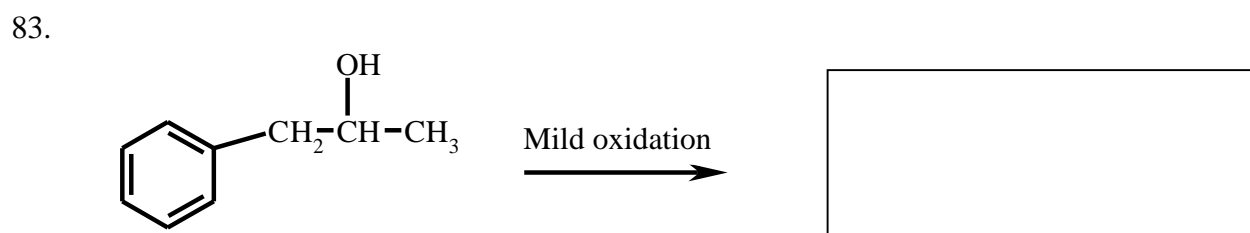
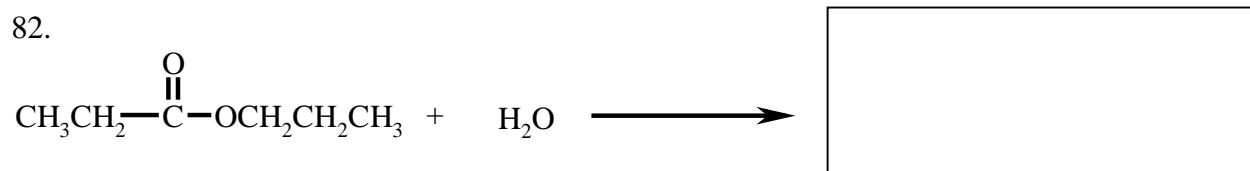
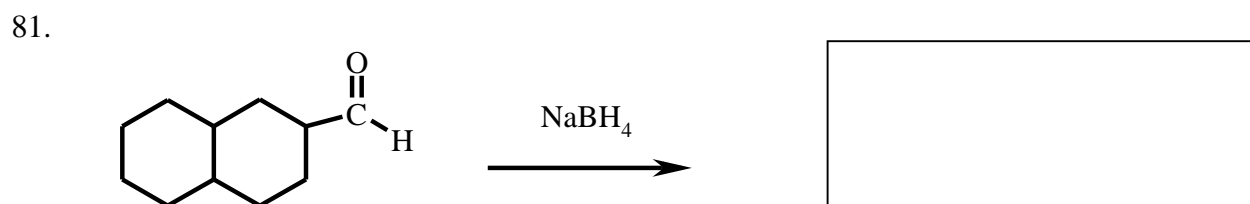
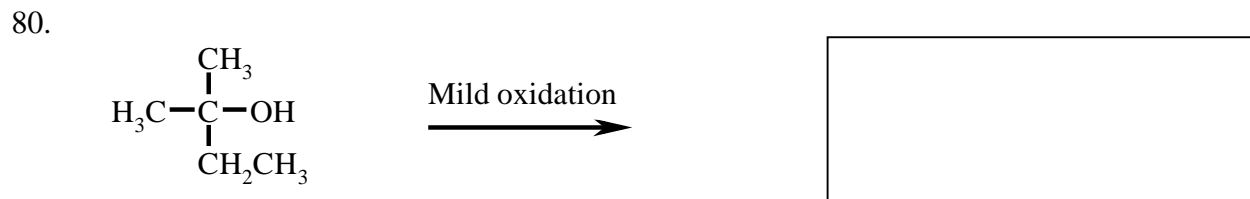
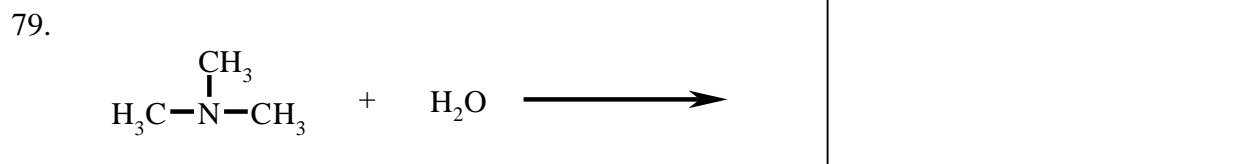
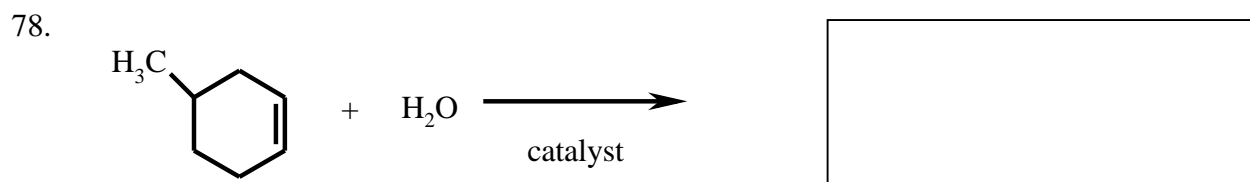
77.



+ H₂

Pd
 \longrightarrow
 catalyst





HAVE A GREAT SUMMER!!!!